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Nitrates/Nitrites and Preservative Replacements in Meat Products

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Nitrates/Nitrites and Preservative Replacements in Meat Products

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OF SCIENCE AND TECHNOLOGY

Clean Label Product Innovation Short Course
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Objectives

- To review the basic functions and importance of nitrites and nitrates as curing agents in meat products
- To present and discuss the various types of “clean label” processed meat products and their regulatory requirements
- To discuss alternatives to replace conventional nitrates/nitrites and antimicrobial preservatives in clean label processed meat products in the context of the technical challenges involved

Agenda

- History of Meat Curing
- Nitrate and Nitrite – The Basics
- Nitrate and Nitrite in the Diet
- Nitrate and Nitrite Content of Foods
- Functions of Nitrate and Nitrite in Meat Products
- Alternative Curing Systems
- Alternative Antimicrobial Systems
- Labeling of Clean/Simple Label Processed Meat Products

History of Meat Curing

- One of the oldest forms of food preservation
- Likely discovered by accident via contamination of salt with saltpeter (potassium nitrate)
- By 10th century, Romans were adding saltpeter intentionally to obtain distinctive color and flavor
- First USDA regulations for nitrite and nitrite use in meat products established in 1925
- Regulations last updated in 1970
- Bacon-specific regulations established in 1978 (to minimize potential for nitrosamine formation)

Nitrate and Nitrite

- Ionic forms:
 - Nitrate – NO_3^-
 - Nitrite – NO_2^-
- Sodium or potassium salts
 - NaNO_3 or KNO_3
 - NaNO_2 or KNO_2

Sodium Nitrate

- Source:
 - Can be mined or synthesized
- In meat products, functions as a source and reservoir of nitrite

Sodium Nitrite

- Source:
 - Commercially synthesized by treating sodium hydroxide with mixture of nitrogen dioxide and nitric oxide:
$$2 \text{ NaOH} + \text{NO}_2 + \text{NO} \rightarrow 2 \text{ NaNO}_2 + \text{H}_2\text{O}$$
- By definition, cured meats must contain nitrite
- Important public health function due to proven track record of *C. botulinum* prevention, which has been further validated by recent research

Nitrate and Nitrite in the Diet

- Two major sources in the body:
 - **Endogenous** → L-arginine-NO synthase pathway
 - NO (nitric oxide) generated by conversion of L-arginine to L-citrulline by nitric oxide synthase (NOS) is oxidized in blood and tissues to form NO_3^- and NO_2^-
 - **Exogenous** → Diet

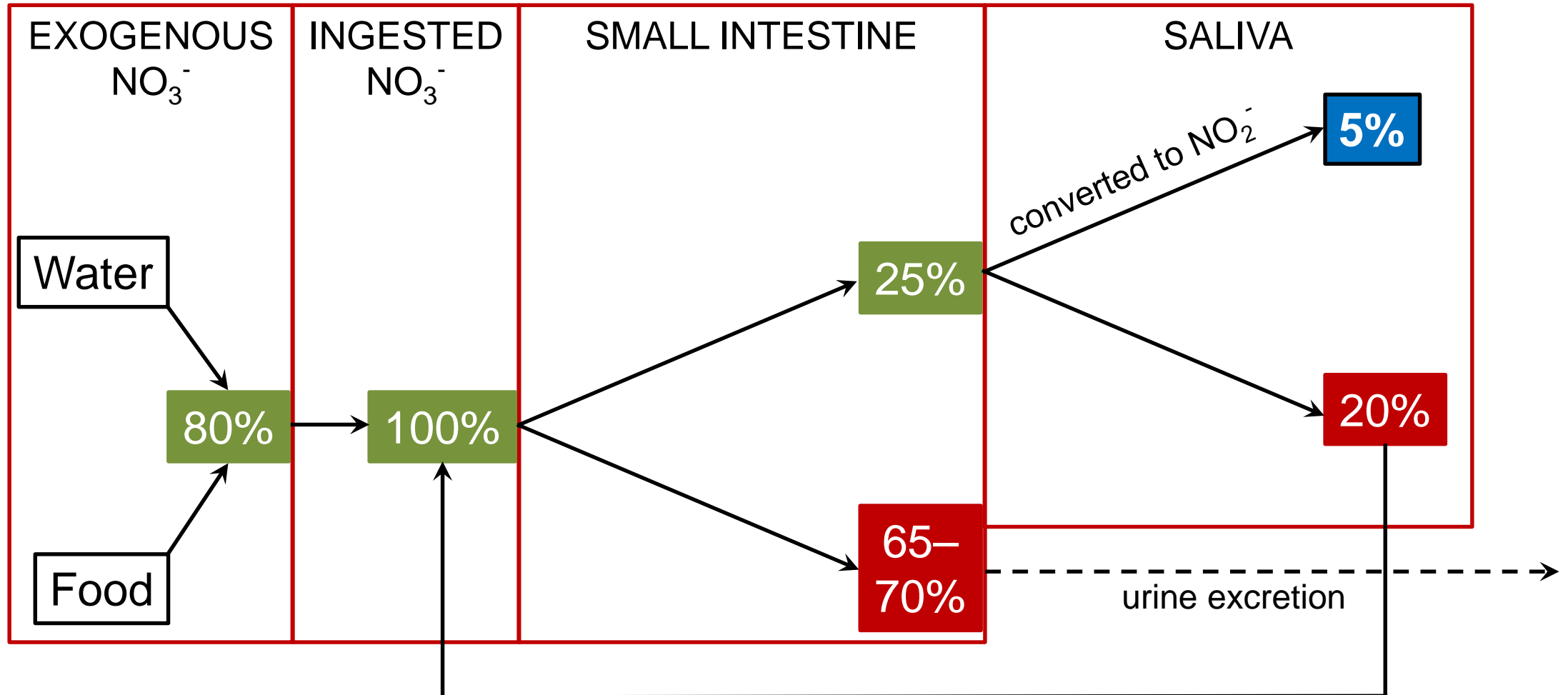
Nitrate in the Diet

- Mean daily intake = 43–141 mg (WHO, 2007)
- Total (exogenous) intake sources
 - Plant foods: $\approx 80\%$
 - Drinking water: 10–15%
- Food intake sources
 - Various vegetables
 - Largest contributors of dietary intake of nitrate; $\approx 85\text{--}90\%$
 - E.g., lettuce, cabbage, spinach, beets, celery, radishes

Nitrite in the Diet

- Mean daily intake = 1.2–3.0 mg (WHO, 2007)
- Total intake sources
 - Saliva: $\approx 93\%$
 - Vegetables: $\approx 2\%$
 - Cured meats: $\approx 5\%$
- Enterosalivary circulation of dietary nitrite
 - Ingested nitrate is absorbed in small intestine, 25% of which is secreted in saliva, $\approx 20\%$ of which is converted to nitrite by oral cavity bacteria
 - Therefore $\approx 5\%$ of ingested nitrate becomes converted to nitrite
 - » This amount accounts for $\approx 93\%$ of total ingested nitrite
 - Nitrite in saliva may produce NO under acidic conditions (as in stomach) in sufficient amounts to protect against swallowed pathogens (Duncan et al. 1995).

Enterosalivary Circulation of Dietary Nitrate



Nitrite/nitrate concentrations of selected foods collected from five U.S. cities in 2009

	Nitrate (ppm)		Nitrite (ppm)	
	Mean	Range	Mean	Range
Broccoli	394	29–1140	0.6	0.01–9.5
Cabbage	418	37–1831	0.1	0.01–0.4
Celery	1495	20–4269	0.1	0.02–0.5
Lettuce	850	79–2171	0.6	0.01–9.7
Spinach	2797	65–8000	8.0	0–137
Cured, dried, uncooked sausage	113	0.1–2289	0.8	0.03–9.7
Cured, cooked sausage	32	0.8–541	8.0	0.1–29.3
Fermented, cooked sausage	46	1.8–320	0.8	0–26.7
Whole-muscle, brine cured, uncooked	14	3.5–32	6.8	0.2–36.5
Whole-muscle, brine cured, cooked	16	0.2–108	7.5	0–27.6
Whole-muscle, dry-cured, uncooked	106	0.4–1367	1.5	0.02–16.2

Source: Keeton JT et al. 2009.

Maximum ingoing nitrite and nitrate limits (in ppm) for meat and poultry products

Curing Agent	Curing Method			
	Immersion Pumped	Massaged or Pumped	Comminuted	Dry Cured
Sodium Nitrite	200	200	156	625
Potassium Nitrite	200	200	156	625
Sodium Nitrate	700	700	1718	2187
Potassium Nitrate	700	700	1718	2187

Source: USDA-FSIS Processing Inspectors' Calculations Handbook. 1995 revision.

<http://www.fsis.usda.gov/OPPDE/rdad/FSISDirectives/7620-3.pdf>.

Potassium nitrate 5%

= 5,000 ppm

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Sensitive

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8 Benefits

- ✓ Sensitivity
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- ✓ Tartar
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- ✓ Freshens
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NET WT 6.0 OZ (170 g)

Drug Facts

Active ingredients

Potassium nitrate 5%

Sodium fluoride 0.24% (0.14% w/v fluoride ion)

Purpose

Antisensitivity
Anticavity

Uses

- builds increasing protection against painful sensitivity of the teeth to cold, heat, acids, sweets or contact
- helps protect against cavities

Warnings

When using this product, if pain/sensitivity still persists after 4 weeks of use, please visit your dentist.

Stop use and ask a dentist if the problem persists or worsens. Sensitive teeth may indicate a serious problem that may need prompt care by a dentist.

Keep out of reach of children. If more than used for brushing is accidentally swallowed, get medical help or contact a Poison Control Center right away.

Drug Facts (continued)

Directions

adults and children 12 years of age and older

apply at least a 1-inch strip of the product onto a soft bristle toothbrush. Brush teeth thoroughly for at least 1 minute twice a day (morning and evening) or as recommended by a dentist or physician. Make sure to brush all sensitive areas of the teeth.

children under 12 years

consult a dentist or physician

Inactive ingredients

water, hydrated silica, glycerin, sorbitol, PEG-12, tetrapotassium pyrophosphate, PVM/MA copolymer, flavor, sodium lauryl sulfate, poloxamer 407, sodium hydroxide, sodium saccharin, cellulose gum, xanthan gum, titanium dioxide

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* Maximum strength FDA allowed antisensitivity active ingredient
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Nerves
Dentin
Exposed Dentin

9 35000 52101 0

Concerns About Nitrite

■ Toxicity

- Very reactive; toxic at very high doses
 - Lethal dose ≈ 1 g (14 mg/kg body weight)
 - » Assuming 15 ppm residual nitrite, this would be equivalent to ingesting 1,086 56-g hot dogs (66.7 kg)
- Concern for methemoglobinemia, primarily in infants (blue baby syndrome)

■ Nitrosamines and Cancer

- N-nitrosamines cancer-causing potential discovered during 1950s/1960s
- Formed by reaction of nitrite with secondary amines near neutral pH and temperatures $>130^{\circ}\text{C}$
- For more information, consult Sindelar & Milkowski (2012)

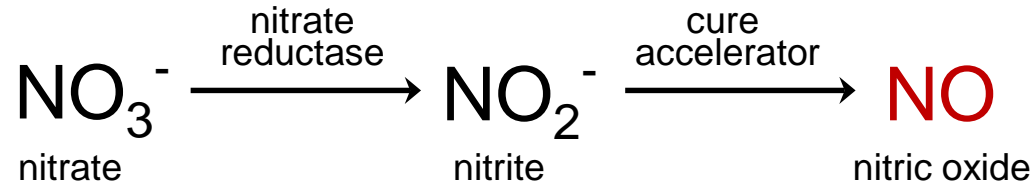
Functions of Nitrite

- **Cured Color**
 - **Cured Flavor**
 - **Texture**
 - **Antimicrobial**
 - **Antioxidant**
- } Organoleptic effects

Functions of Nitrite

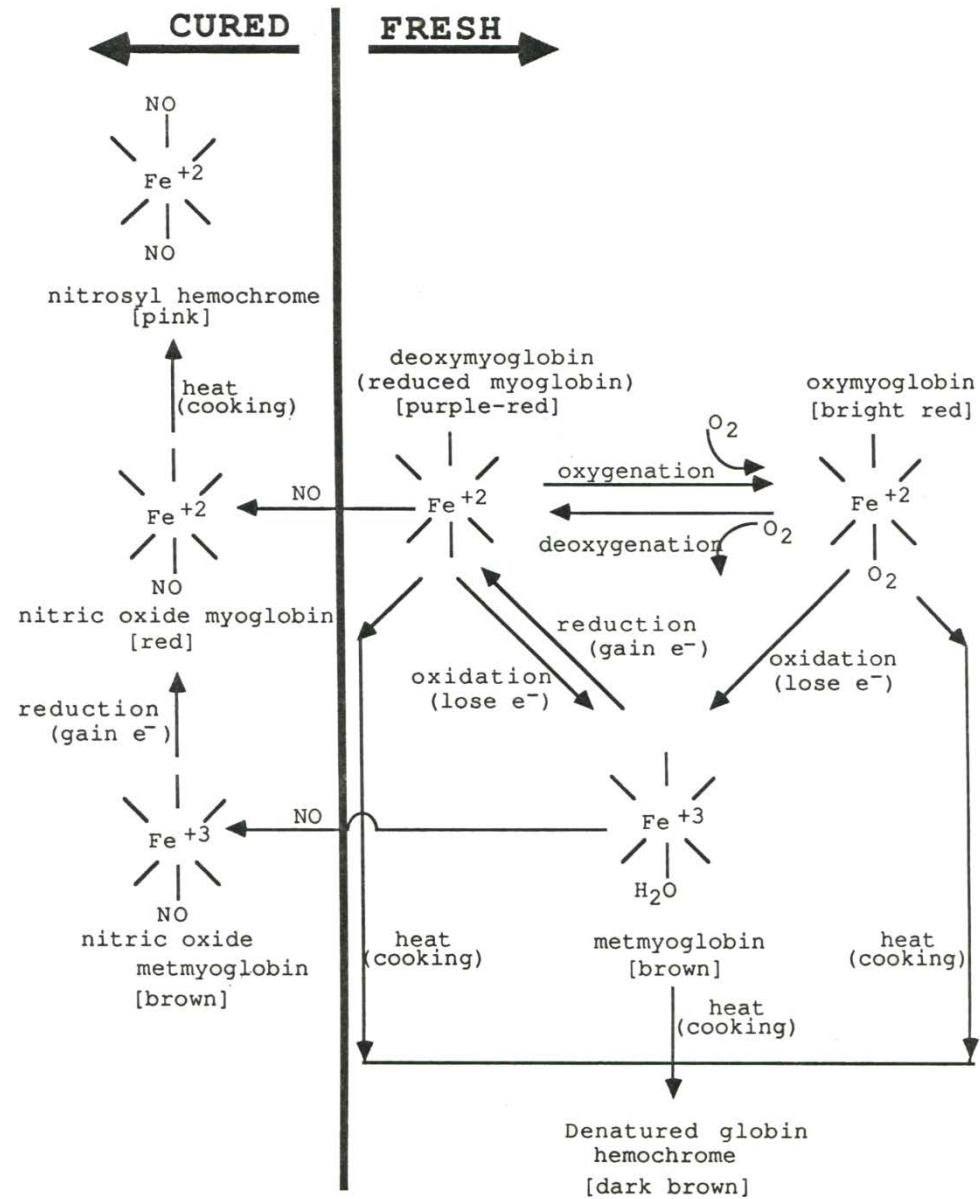
1. Color

- Due to nitric oxide (NO) production from NO_2^-



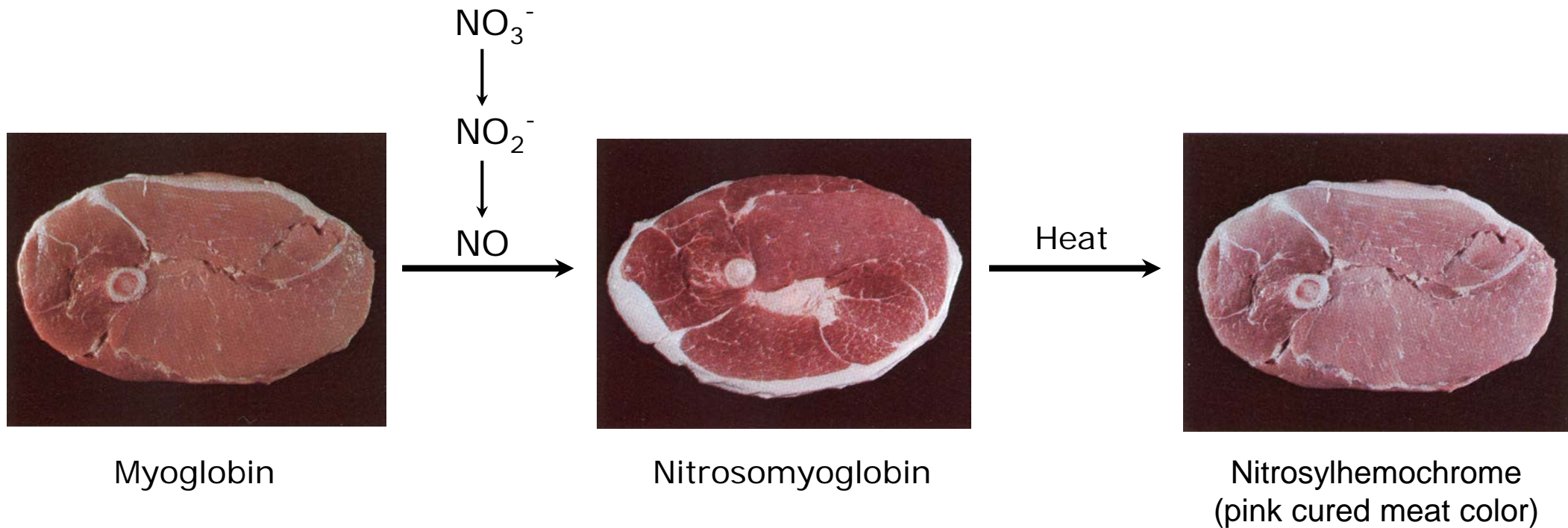
- <5 ppm NO_2^- required for this effect
- Promoted by acidic conditions
- Accelerated by reductants, salt, pigments (Fe)

MEAT COLOR FORMS



Functions of Nitrite

- Interaction of NO with meat pigment myoglobin



Functions of Nitrite

2. Antimicrobial

- Inhibitory activity towards many microorganisms, including *Clostridium botulinum*
- Acts in synergism with salt and other antimicrobials, e.g., lactate, to control *Listeria monocytogenes*

Functions of Nitrite

3. Antioxidant

- Effective for preventing rancidity
 - Decreases rate of lipid oxidation
 - Eliminates “warmed-over flavor” in cooked meats
 - Interacts with heme iron
 - NO scavenges free radicals
- Antioxidant activity of NO_2^- may play a role in cured flavor

Functions of Nitrite

4. Flavor

- Cured flavor not clearly understood
- Is it a specific flavor compound or *the absence* of other flavor compounds? Is it both?
 - Fewer flavor compounds have been isolated from cooked cured meats than from uncured cooked meats
 - Does antioxidant activity play a role?

5. Texture

- Protein crosslinking of sulfur-containing amino acids by nitrite

Functions of Nitrite

No other known substance possesses all of
nitrite's unique properties

Functions of Nitrate

- Serves as nitrite precursor and/or reservoir
- Sometimes included in starter cultures as a cured color promoting agent
- Also available from natural sources, such as celery powder and swiss chard
 - “Natural” or “Alternative” curing
 - Requires conversion to nitrite
 - In-plant fermentation
 - Pre-converted

Nitrate/Nitrite Alternatives

- Health concerns over nitrite in the 1960s lead to development of truly uncured (nitrite-free) products in the 1970s
 - These provided inferior organoleptic, safety and shelf-life properties than their cured counterparts
- In response, in 1972 USDA established requirement that products “for which there is a standard and to which nitrate or nitrite is permitted or required to be added, may be prepared without nitrate or nitrite and labeled with such standard name when immediately preceded with the term ‘**Uncured**’...”
- 1979 amendment include statement “**No Nitrate or Nitrite Added.**”

Clean/Simple Label Alternatives in Processed Meats

- Elements to replace:
 - Curing system
 - Nitrate/nitrite, ascorbate/erythorbate
 - Antimicrobial system
 - Lactate/diacetate, benzoate, propionate, others
 - Any ingredient called by its chemical name
 - Other ingredients
 - Phosphates, binders (e.g., carrageenan, *modified* starches)
 - Any ingredient called by its chemical name
- This necessitates search for “natural” or simple-name versions of these ingredients or other ingredients with similar functionality

Alternative Curing Systems

- Three basic elements of an alternative curing system
 - Nitrate/nitrite source
 - Cure accelerator
 - Bacterial starter culture (not always necessary)

Alternative Curing Systems

1. Nitrate/nitrite source

- Commercial vegetable juices and juice powders
 - Celery (most common); Swiss chard
 - Other sources (e.g., lettuce, spinach, beets) not yet exploited commercially for this purpose

Alternative Curing Systems

1. Nitrate/nitrite source, cont.

- Types:
 - Regular
 - » Commercial sources vary from 20,000–30,000 ppm NO_3^-
 - » Typically standardized to a specific NO_3^- content
 - » Require fermentation step during processing to convert NO_3^- to NO_2^-
 - Pre-converted
 - » NO_3^- converted to NO_2^- by supplier
 - » Typically standardized to specific NO_2^- content (up to >15,000 ppm)
 - » Eliminates the need for fermentation step

Alternative Curing Systems

1. Nitrate/nitrite source, cont.

- Should try to maximize usage level to increase ingoing NO_2^- content
- There's no regulatory limit, but, in most cases, self-limiting at 0.3–0.4% due to strong celery flavor
 - Maximum level is formulation specific, with blander-tasting products tolerating less
 - Opportunity for flavor research??

Alternative Curing Systems

1. Nitrate/nitrite source, cont.

- Concern
 - Does system generate enough nitrite to control *C. botulinum*, *C. perfringens* and *L. monocytogenes* ?
 - » E.g., pre-converted celery powder (22,500 ppm NaNO_2) at 0.4% would result in 90 ppm ingoing NaNO_2 . Is that enough?
 - » It has been estimated that at least 70 ppm is needed for adequate antilisterial activity (Glass et al., 2008)
 - » Commercial alternatively-cured frankfurters, hams and bacon have been shown to have more potential for growth of *C. perfringens* (Jackson et al., 2011)
 - » Effectiveness of nitrite is affected by of other factors (e.g., other antimicrobials, salt level, temperature, a_w , etc.)

Alternative Curing Systems

2. Lactic acid starter culture

- Reduce NO_3^- to NO_2^- via production of nitrate reductase
- Commercially available strains
 - *Staphylococcus carnosus*, *S. utilis*, *S. vitulinus*, *Micrococcus varians*
- Requires fermentation step (1–2 h at $\approx 27\text{--}38^\circ\text{C}$) prior to cooking to allow for NO_2^- formation
 - Conversion not 100% efficient; difficult to control $[\text{NO}_2^-]$
- Not necessary when using pre-converted NO_2^- sources

Alternative Curing Systems

3.Cure Accelerators

- Natural sources of ascorbic acid
- Types
 - Cherry powder
 - Acerola powder
 - Lemon juice powder (to lower pH)

Antimicrobial Systems

■ Conventional

- Nitrite
- Lactate
- Diacetate
- Benzoate
- Propionate

■ Simple Name and or Natural

- Cultured sugars
- Vinegar / buffered vinegar
- Lemon juice / lemon juice solids
- Plant extracts (e.g., rosemary, oregano, sage)

- Much commercial development activity in this area
- Need to confirm antimicrobial effectiveness (safety and shelf-life) on case-by-case basis by use of microbial challenge studies
- Some commercial microbial growth predictive models incorporate simple name antimicrobials

Labeling

1. Uncured Products

- “Any product, such as frankfurters and corned beef, for which there is a standard in this part and to which nitrate or nitrite is permitted or required to be added, may be prepared without nitrate or nitrite and labeled with such standard name when immediately preceded with the term “**Uncured**” in the same size and style of lettering as the rest of such standard name: *Provided*, That the product is found by the Administrator to be similar in size, flavor, consistency, and general appearance to such product as commonly prepared with nitrate and nitrite...” 9 CFR 319.2

Labeling

1. Uncured Products, cont.

- Products “which contain no nitrate or nitrite shall bear the statement “**No Nitrate or Nitrite Added.**” This statement shall be adjacent to the product name in lettering of easily readable style and at least one-half the size of the product name.” 9 CFR 317.17(c)(1)
- These products must also bear the statement “**Not Preserved—Keep Refrigerated Below 40°F At All Times**” unless they have been thermally processed to F_0 of 3 or more; they have been fermented or pickled to pH of 4.6 or less; or they have been dried to a water activity of 0.92 or less.” 9 CFR 317.17(c)(2)

Labeling



Labeling

2. “Natural” Products

- USDA-FSIS Food Standards and Labeling Policy Book:
 - The term “natural” may be used on labeling for meat products and poultry products, provided the applicant for such labeling demonstrates that: (1) the product does not contain any artificial flavor or flavoring, coloring ingredient, or chemical preservative (as defined in 21 CFR 101.22), or any other artificial or synthetic ingredient; and (2) the product and its ingredients are not more than minimally processed.
 - Therefore, sodium nitrate or nitrite are not allowed
 - » Products must be labeled “uncured”

Labeling

2. “Natural” Products, cont.

- USDA-FSIS Food Standards and Labeling Policy Book:
 - Minimal processing may include: (a) those traditional processes used to make food edible or to preserve it or to make it safe for human consumption, e.g., smoking, roasting, freezing, drying, and fermenting, or (b) those physical processes which do not fundamentally alter the raw product and/or which only separate a whole, intact food into component parts, e.g., grinding meat, separating eggs into albumen and yolk, and pressing fruits to produce juices.

Labeling



Labeling

3. Organic Products

- Governed by Organic Foods Production Act (1990) and managed by the National Organic Program (NOP)
 - Establishes prohibited practices, requirements and allowed substances
- Synthetic and non-organic substances that may be used are listed in the National List of Allowed and Prohibited Substances (7 CFR 205.605 and 606)
- Sodium nitrate or nitrite specifically prohibited
 - Products must, therefore, be labeled “uncured.”

Labeling

3. Organic Products, cont.

- Regulated by Organic Foods Production Act (1990):
- “100% Organic”
 - Only organically-produced ingredients and processing aids
 - Water and salt are excluded
 - Can use USDA organic seal
- “Organic”
 - 95%+ organically-produced ingredients, excluding water and salt
 - Can use USDA organic seal
- “Made with Organic Ingredients”
 - 70+ organically-produced ingredients
 - Cannot use USDA organic seal

Labeling



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Thank You



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